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**TiCl(OCH<sub>3</sub>)(SO<sub>3</sub>Cl)(OSO<sub>2</sub>F)-CHLORO(CHLOROSULFATO)(FLUOROSULFATO)-  
METHOXYTITANIUM(IV) – ASYMMETRIC TITANIUM(IV) COMPOUND**

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A stable titanium(IV) compound with four different kinds of ligands, TiCl(OCH<sub>3</sub>)(SO<sub>3</sub>Cl)-(SO<sub>3</sub>F), results when TiCl<sub>2</sub>(OCH<sub>3</sub>)(SO<sub>3</sub>F) is refluxed with HSO<sub>3</sub>Cl using CH<sub>2</sub>Cl<sub>2</sub> as the solvent. It is an off-white solid and decomposes around 125°C. The powder X-ray diffraction studies of the title compound will be discussed along with other physical and chemical properties.

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**CRYSTAL STRUCTURE OF: FLUOSILICATE TETRAHYDRATE OF LEAD**

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The crystal structure of Pb SiF<sub>6</sub> · 4H<sub>2</sub>O, first observed by Mariqnac (1860) has been solved and refined from single crystal X-ray data. The compound crystallizes in the monoclinic space group P 2<sub>1</sub>/c with the following cell parameters : a = 7.839(1) ; b = 7.998(2) ; c = 12.650(2) Å ; β = 91.54(1)° Z = 4 ; ρ<sub>calc</sub> = 3.530(5) Mg m<sup>-3</sup>.

This structure is a new one among fluosilicate hydrates of divalent metals with a marked unidimensionnal character. The metal has a mixed environment of fluorine atoms and water molecules in a bicapped square antiprism arrangement.